Table 1. Experimental parameters used to measure the electrochemical impedance response of a three-electrode MEA fuel cell

Applied E vs. Interposed Pt Membrane (V)	AC Perturbation Amplitude (mV)	Resistive Load (Ω)	Thermodynamic Effect
Open Circuit <0.448>	10	5	Cathodic polarization, pseudo-equilibrium condition of cell under load
0.708	10	5	Anodic polarization counteractive to oxygen reduction at cathode
1.000	10	5	Anodic polarization, very counteractive to oxygen reduction at cathode
-0.628	10	5	Cathodic polarization, conducive to oxygen reduction at cathode
0.628	10	5	Anodic polarization, slightly counteractive to oxygen reduction at cathode

Table 2 Results of model parameter fits as a function of the applied polarization potential (thermodynamic state) at the cathode.

Applied E vs.	Measured	Pore Electrolyte	Charge Transfer	Cell	Cell	
Interposed Pt	Current	Resistance, Z <sub>pore</sub>	Resistance, R <sub>ct</sub>	Resistance	Capacitance	Distribution
Membrane (V)	(μA)	(Ω)	(Ω)	$(\Omega)$	(mF)	Exponent, β
OCP	-102	11.9	88.1	33.6	1.78	0.934
<0.448>		±17.4%	±12.3%	±0.85%	±5.21%	±0.31%
0.708	213	1.96	110	10.5	0.233	0.784
		±19.1 %	±13.3%	±0.54%	±3.48%	±0.22%
1.000	860	→0†	1094	25.4	0.00117	0.684
			±4.36%	±13.2%	±7.37%	±0.26%
-0.628	-50,026	1.66	0.827	7.93	7.65	0.622
0.628	-1.14	8.70	61.9	10.2	1.29	0.937
		±14.8%	±13.2%	±15.8%	±5.83%	±0.43%

<sup>†</sup> Dominated by charge transfer resistance (R<sub>ct</sub>)